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Phytochemistry 63 (2003) 445–448

PHYTOCHEMISTRY

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Two prenylated flavonoids from the stem bark of *Erythrina burttii*

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Received 7 November 2002; received in revised form 12 March 2003

Abstract

From the stem bark of *Erythrina burttii*, a new isoflavone, 5,2',4'-trihydroxy-7-methoxy-6-(3-methylbut-2-enyl)isoflavone (trivial name, 7-*O*-methylluteone) and a new flavanone, 5,7-dihydroxy-4'-methoxy-3'-(3-methylbutadienyl)-5'-(3-methylbut-2-enyl)-flavanone (trivial name, burttinonedehydrate) along with three known isoflavonoids (8-prenylluteone, 3-*O*-methylcalopocarpin and genistein) were isolated. The structures were determined on the basis of spectroscopic evidence.

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Keywords: *Erythrina burttii*; Leguminosae; Stem bark; Isoflavone; Flavanone; 7-*O*-Methylluteone; Burttinonedehydrate

1. Introduction

The genus *Erythrina* (Leguminosae) is widely used for treatment of microbial infections in traditional medicinal practice (Mitscher et al., 1987). This genus is a rich source of prenylated flavonoids and isoflavonoids. Some of the flavanones, isoflavones, isoflavanones and pterocarpans isolated from this genus have been shown to possess significant anti-microbial activities (Kamat et al., 1981; Mitscher et al., 1987, 1988). Pterocarpans and isoflavanones with anti-HIV activities have also been reported from two *Erythrina* species (McKee et al., 1997).

In earlier work on *Erythrina* species of Kenya, we have reported new flavonoids and isoflavonoids from the stem bark (Yenesew et al., 1998b) and the root bark (Yenesew et al., 2002) of *Erythrina burttii* and also from the stem bark of *E. saculeuxii* (Yenesew et al., 1998a; Yenesew et al., 2000). We report here a new isoflavone (**1**) and a new flavanone (**2**) along with three known isoflavonoids from the stem bark of *Erythrina burttii*.

2. Results and discussion

Compound **1** was isolated as amorphous solid. The UV (λ_{\max} 266, 289sh and 331 nm), ¹H (δ 8.09, s, for H-2) and ¹³C (δ 154.8 for C-2, 123.3 for C-3 and 181.9 for C-4) NMR spectra were consistent with an isoflavone skeleton. In addition, the presence of three hydroxy (one of which is chelated), a methoxy and a 3,3-dimethylallyl substituents were evident from the MS (M^+ 368, C₂₁H₂₀O₆) and NMR data (Table 1). In the MS, the presence of a fragment ion at m/z 134, resulting from a retro-Diels–Alder cleavage of C-ring, would place the two hydroxy groups in B-ring, and hence A-ring should contain the chelated hydroxy (at C-5), the methoxy and the 3-methylbut-2-enyl groups. In the NOESY spectrum, NOE interactions of the methoxy with the aromatic proton at δ 6.48 (s) and the methylene protons of the 3,3-dimethylallyl group (δ 3.38), are in agreement with the placement of the methoxy group at C-7. Then the 3-methylbut-2-enyl group could either be at C-6 or C-8. The attachment of this group at C-6 was deduced from the HMBC experiment, which showed correlation of the methylene protons of the 3-methylbut-2-enyl group with C-5 and C-7, and also correlation of H-8 with C-7 and C-8a (Table 1).

In the B-ring, the occurrence of an AX₂ spin system (δ 7.01, *d*, $J=8.1$ Hz for H-6'; 6.47, *dd*, $J=1.5, 8.1$ Hz

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Table 1
¹H (CDCl₃ at 300 MHz) and ¹³C (CDCl₃ at 75 MHz) NMR chemical shift data, together with HMBC correlations for **1**

	¹ H (<i>J</i>)	¹³ C	² <i>J</i>	³ <i>J</i>
2	8.09 <i>s</i>	154.8	C-3	C-1', C-4, C-4a
3		123.3		
4		181.9		
4a		105.6		
5		158.1 ^a		
6		113.8		
7		164.1		
8	6.48 <i>s</i>	89.5	C-7, C-8a	C-6, C-4a
8a		156.4		
1'		112.5		
2'		157.3		
3'	6.55 <i>d</i> (1.5 Hz)	106.4	C-2', C-4'	C-1', C-5'
4'		158.3 ^a		
5'	6.47 <i>dd</i> (1.5, 8.1 Hz)	108.6	C-4'	C-1', C-3'
6'	7.01 <i>d</i> (8.1 Hz)	130.7		
1''	3.38 <i>d</i> (6.9 Hz)	21.5	C-6, C-2''	C-5, C-7, C-3''
2''	5.20 <i>t</i> (6.9 Hz)	121.5		C-4'', C-5''
3''		132.3		
4''	1.87 <i>s</i>	17.8	C-4''	C-2''
5''	1.68 <i>s</i>	25.8	C-5''	C-2''
OMe-7	3.93 <i>s</i>	56.1		C-7
OH-5	12.36 <i>s</i>			

^a May be interchangeable.

and 6.55, *d*, *J* = 1.5 Hz) for the aromatic protons places the two hydroxy groups in this ring at C-2' and-4'. Also, the chemical shift values of the carbon atoms in this ring are in agreement with such oxygenation pattern (Nkengfack et al., 1989; Taylor et al., 1986). On these grounds this new compound was identified as 5,2',4'-trihydroxy-7-methoxy-6-(3-methylbut-2-enyl)isoflavone (**1**) for which the trivial name 7-*O*-methylluteone is suggested.

EIMS analysis of the second compound (**2**) isolated from the stem bark of this plant showed a molecular ion peak at *m/z* 420 (C₂₆H₂₈O₅). The UV, ¹H and ¹³C NMR data (see Section 3) of this new compound suggested that it should be a flavanone derivative. Furthermore the MS (*m/z* 153 resulting from RDA cleavage of C-ring) and NMR spectra suggested that compound **2** should have identical A-ring as those of abyssinone V-4'-methyl ether (**3**) and burttinone (**4**) which were earlier reported from the stem bark of this plant (Yenesew et al., 1998b). As in **3** and **4**, B-ring is tri-substituted at C-3',-4' and-5' with one methoxy and two C₅ groups. The chemical shift value (δ 61.7) of the methoxy group in ¹³C NMR is typical of a *di-ortho* substituted methoxy position. Thus the methoxy group should be placed at C-4', leaving the two C₅ units at C-3' and C-5'. One of the C₅ groups is readily identified as a 3-methylbut-2-enyl group from NMR (see Section 3), while the other should be a 3-methylbutadienyl, as deduced from ¹H (a pair of doublets at δ 6.84 and 6.40, *J* = 16.2 Hz, for H-1'' and 2'', a singlet at 1.83 for Me-4'' and a broad singlet at

5.00 for CH₂-5'') and ¹³C (122.7 for C-1'', 133.6 for C-2'', 142.6 for C-3'', 24.6 for C-4'' and 114.1 for C-5'') NMR. This compound was hydrogenated over Pd/C to give **2a**. Similar hydrogenation of **3** (isolated and characterized from the stem bark of *E. burttii* as described in Yenesew et al., 1998b) also gave **2a**. From the negative optical rotation, [α]_D –66°, the absolute stereochemistry at C-2 was assumed to be *S*, as the other flavanones of this plant (Promsattha et al., 1986). Thus this compound was characterized as (*S*)-5,7-dihydroxy-4'-methoxy-3'-(3-methylbutadienyl)-5'-(3-methylbut-2-enyl)flavanone, for which the trivial name burttinonedehydrate (**2**) is given by relating it to its likely precursor burttinone (**4**).

From stem bark of this plant three known isoflavonoids were also isolated and identified as 8-prenylluteone (Nkengfack et al., 1989), 3-*O*-methylcalopocarpin (McKee et al., 1997) and genistein.

3. Experimental

3.1. General

Analytical TLC: Merck pre-coated silica gel 60 F₂₅₄ plates. CC on silica gel 60 (70–230 mesh). EIMS: direct inlet, 70 eV, on a SSQ 710, Finnigan MAT mass spectrometer. ¹H and ¹³C NMR on ARX 300 (Bruker) or Varian-Mercury spectrometers using TMS as int. standard. HMQC and HMBC spectra were acquired using the standard Bruker software.

3.2. Plant material

The stem bark of *Erythrina burttii* Ball. f. was collected near Emali town, on the Nairobi-Mombasa road, Kenya, in February 2001. The plant was identified at the University Herbarium, Botany Department, University of Nairobi, where a voucher specimen is deposited.

3.3. Extraction and isolation

Dried and ground stem bark (2 kg) of *E. burttii* was extracted with acetone by cold percolation. Removal of the solvent afforded a brown gummy extract (110 g). A 100-g portion of the extract was subjected to CC on silica gel (500 g) eluting with hexane containing increasing amounts (3, 5, 10, 15, 20, 30, 50%) of EtOAc. A total of seven fractions each of ca. 2 l were collected. Crystallization of the fraction eluted with 3% EtOAc in hexane gave 1.5 g of **3** (Yenesew et al., 1998b). The mother liquor of this fraction was further subjected to CC on silica gel eluting with hexane/acetone (5:1) to give **2** (22 mg), 3-*O*-methylcalopocarpin (16 mg) and a further amount of **3** (247 mg). The fraction eluted with 15% EtOAc was further purified by CC on Sephadex

LH-20 (eluted with $\text{CH}_2\text{Cl}_2/\text{MeOH}$; 1:1) and PTLC ($\text{CH}_2\text{Cl}_2/\text{MeOH}$; 20:1) to give 8-prenylluteone (28 mg), while similar treatment of the fraction eluted with 20% EtOAc gave **1** (32 mg) and genistein (13 mg).

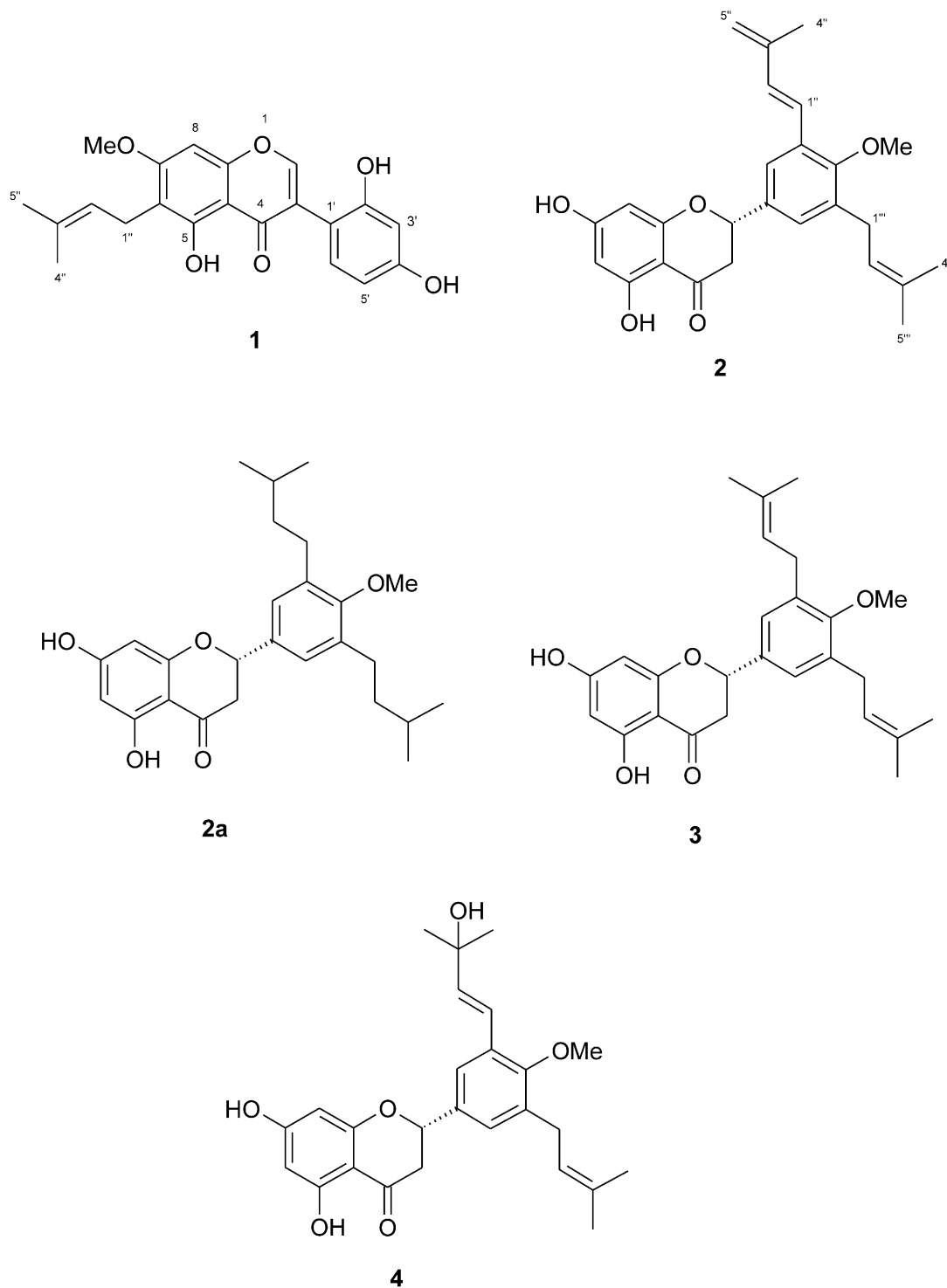
3.4. 7-O-Methyllyuteone (**1**)

Amorphous solid. UV λ_{max} (MeOH) nm: 212, 266, 289sh, 331. ^1H NMR (Table 1). ^{13}C NMR (Table 1).

EIMS m/z (rel. int.): 368 (17, $[\text{M}]^+$), 325 (55), 313 (89), 235 (8), 191 (17), 179 (100), 149 (66), 134 (65).

3.5. Burttinonedehydrate (**2**)

Oil. $[\alpha]_{\text{D}} -66^\circ$ (MeOH, c 0.01). UV λ_{max} (MeOH) nm: 288, 330. ^1H NMR (acetone- d_6 , 200 MHz): δ 5.28 (1H, dd , $J=2.7, 13.2$ Hz, H-2), 2.71 (1H, dd , $J=2.7, -17.0$ Hz, H-3 α), 3.04 (1H, dd , $J=13.2, -17.0$ Hz, H-3 β), 5.98



(2H, br s, H-6 and H-8), 7.72 (1H, d, $J=1.5$ Hz, H-2'), 7.32 (1H, d, $J=1.5$ Hz, H-6'), 6.84 (1H, d, $J=16.2$ Hz, H-1''), 6.40 (1H, d, $J=16.2$ Hz, H-2''), 1.83 (3H, s, Me-4''), 5.00 (2H, br s, H-5''), 3.36 (2H, d, $J=7.2$ Hz, H-1'''), 5.28 (1H, t, $J=7.2$ Hz, H-2'''), 1.72 (6H, s, Me-4''' and Me-5'''), 3.77 (3H, s, OMe), 12.07 (1H, s, 5-OH). ^{13}C NMR (acetone- d_6 , 50 MHz): δ 79.2 (C-2), 43.3 (C-3), 196.3 (C-4), 164.4 (C-5), 97.0 (C-6), 166.6 (C-7), 95.9 (C-8), 163.4 (C-9), 103.5 (C-10), 135.2 (C-1' or C-5'), 122.5 (C-2''), 130.8 (C-3'), 156.3 (C-4'), 135.6 (C-5' or C-1'), 125.0 (C-6'), 122.7 (C-1'' or C-2'''), 133.6 (C-2''), 142.6 (C-3''), 24.6 (Me-4''), 114.1 (C-5''), 28.6 (C-1'''), 122.4 (C-2''' or C-1'''), 133.5 (C-4'''), 18.2 (Me-4'''), 26.0 (Me-5'''), 61.7 (OMe). EIMS m/z (rel. int.): 420 (30, $[\text{M}]^+$), 253 (13), 241 (25), 179 (18), 173 (23), 153 (50), 69 (100).

3.6. Hydrogenation of **2**

Compound **2** (15 mg) was hydrogenated over Pd/C to give **2a** (12 mg) as oil. ^1H NMR (acetone- d_6 , 200 MHz): δ 5.41 (1H, dd, $J=2.7, 13.2$ Hz, H-2), 2.68 (1H, dd, $J=2.7, 17.0$ Hz, H-3 α), 3.16 (1H, dd, $J=13.2, -17.0$ Hz, H-3 β), 5.97 (2H, br s, H-6 and H-8), 7.24 (2H, s, H-2' and H-6'), 2.66 (4H, m, CH₂-1'' and CH₂-1'''), 1.57 (4H, m, CH₂-2'' and CH₂-2'''), 1.60 (2H, m, H-3'' and H-3'''), 0.96 (12H, d, $J=6.4$ Hz, Me-4'', Me-5'', Me-4''' and Me-5'''), 3.75 (3H, s, OMe), 12.20 (1H, s, 5-OH), 9.98 (1H, s, OH-7). ^{13}C NMR (acetone- d_6 , 50 MHz): δ 79.5 (C-2), 43.1 (C-3), 196.5 (C-4), 164.7 (C-5), 96.3 (C-6), 166.9 (C-7), 95.3 (C-8), 163.7 (C-9), 102.6 (C-10), 134.7 (C-1'), 126.7 (C-2' and 6'), 136.4 (C-3' and C-5'), 157.0 (C-4'), 28.3 (C-1'' and C-1'''), 28.0 (C-2'' and C-2'''), 40.3 (C-3'' and C-3'''), 22.4 (Me-4'', Me-5'', Me-4''' and Me-5'''), 60.1 (OMe). EIMS m/z (rel. int.): 426 $[\text{M}]^+$ (80), 153 (100). Similarly, hydrogenation of **3** gave **2a**.

Acknowledgements

We acknowledge support by the Deutsche Forschungsgemeinschaft, Germany, Grant No. Pe 264/14-2

and by the Bundesministerium für Zusammenarbeit, Grant No. Pe264/14-3. Mr S. G. Mathenge is thanked for identification of the plant material. S.D. is grateful to the German Academic Exchange Service (DAAD) and the Natural Products Research Network for Eastern and Central Africa (NAPRECA) for a scholarship.

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